

Physics 24100

# **Electricity & Optics**

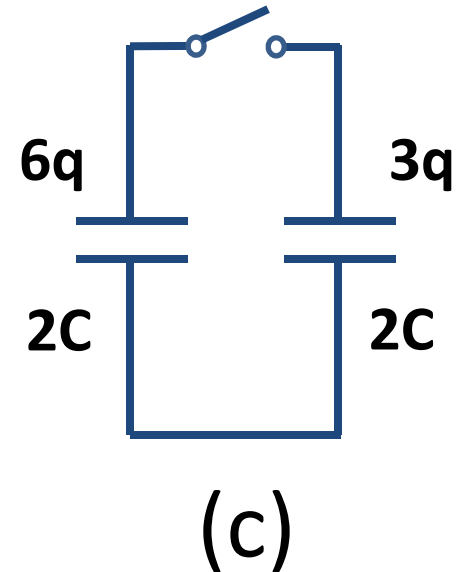
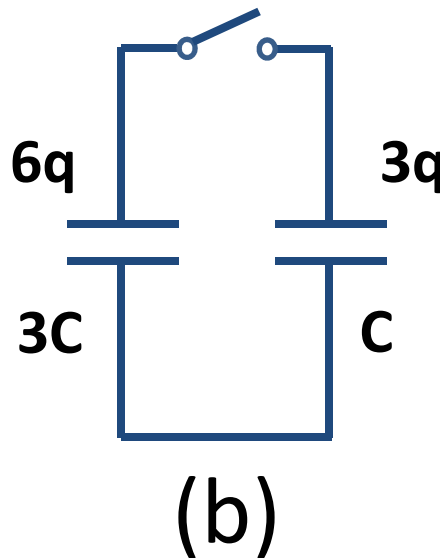
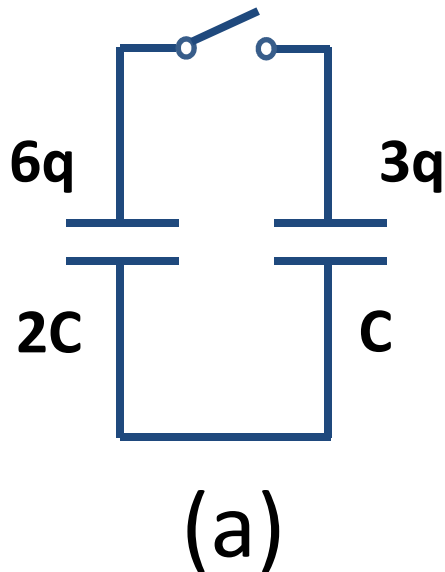
Lecture 10 – Chapter 25 sec. 1-3

Fall 2012 Semester

Matthew Jones

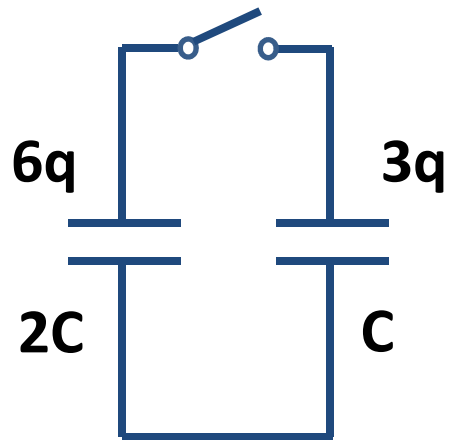
## Tuesday's Question

- Three circuits, consisting of two capacitors and a switch, are initially charged as indicated.
- After the switches are closed, in which circuit will the charge on the **left** *increase*?

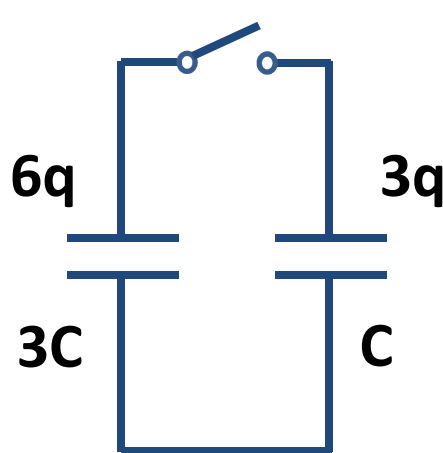


(d) None of them

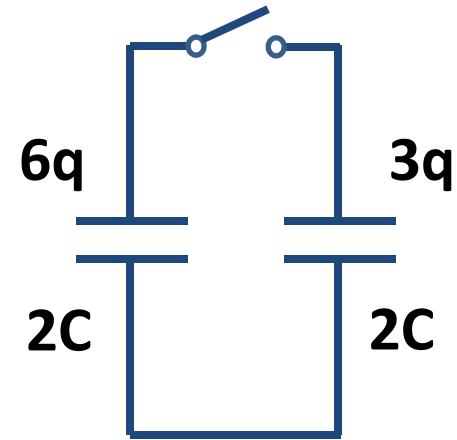
## Tuesday's Question



(a)



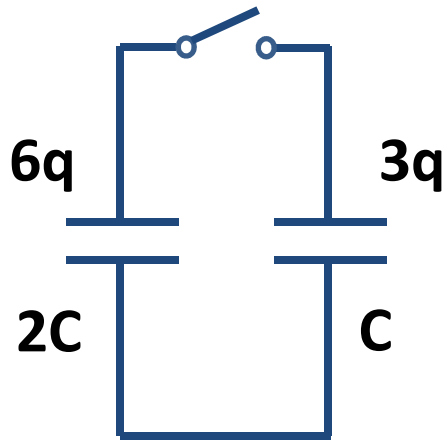
(b)



(c)

- Charge is conserved,  $Q = 9q$
- Calculate equivalent capacitance,  $C_{equiv}$
- Then calculate,  $V = Q/C_{equiv}$
- Finally, calculate,  $Q_{left} = C_{equiv}V$

# Tuesday's Question



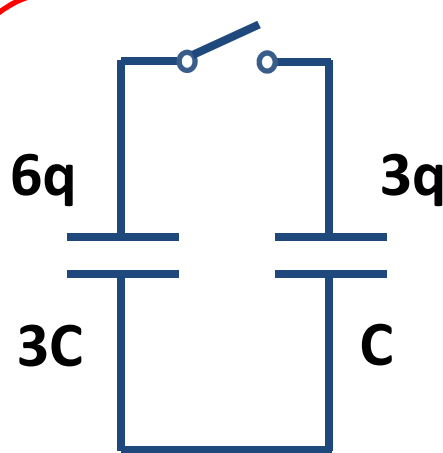
(a)

$$Q = 9q$$

$$C_{equiv} = 3C$$

$$V = \frac{Q}{C_{equiv}} = \frac{3q}{C}$$

$$Q_{left} = 2C \times \frac{3q}{C} = 6q$$



(b)

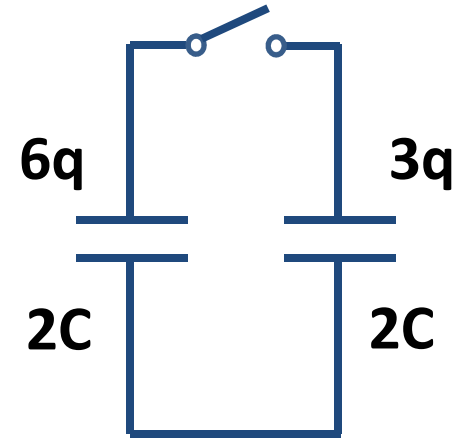
$$Q = 9q$$

$$C_{equiv} = 4C$$

$$V = \frac{Q}{C_{equiv}} = \frac{9q}{4C}$$

$$Q_{left} = 3C \times \frac{9q}{4C}$$

$$= \frac{27}{4}q > 6q$$



(c)

$$Q = 9q$$

$$C_{equiv} = 4C$$

$$V = \frac{Q}{C_{equiv}} = \frac{9q}{4C}$$

$$Q_{left} = 2C \times \frac{9q}{4C} = \frac{9}{2}q$$

$$< 6q$$

# Mini-Review

- Lecture 1:  $\vec{F} = \frac{1}{4\pi\epsilon_0} \frac{Q_1 Q_2}{r^2} \hat{r}$
- Lecture 2:  $\vec{E} = \frac{1}{4\pi\epsilon_0} \frac{Q}{r^2} \hat{r}$  and  $\vec{F} = q\vec{E}$
- Lecture 3:  $\Delta\vec{E}(\vec{x}_p) = \frac{1}{4\pi\epsilon_0} \frac{\Delta Q(\vec{x}_s)}{r^2} \hat{r} \rightarrow \vec{E}(\vec{x}_p) = \frac{1}{4\pi\epsilon_0} \int \frac{\hat{r}}{r^2} dQ$
- Lecture 4:  $\phi_{net} = \oint_S \hat{n} \cdot \vec{E} dA = \frac{Q_{inside}}{\epsilon_0}$
- Lecture 5:  $\vec{E}$  near conductors and insulators
- Lecture 6:  $\Delta V = -\int_a^b \vec{E} \cdot d\vec{\ell}$  and  $\vec{E} = -\vec{\nabla}V$   
$$V(r) = \frac{1}{4\pi\epsilon_0} \frac{Q}{r}$$

# Mini-Review

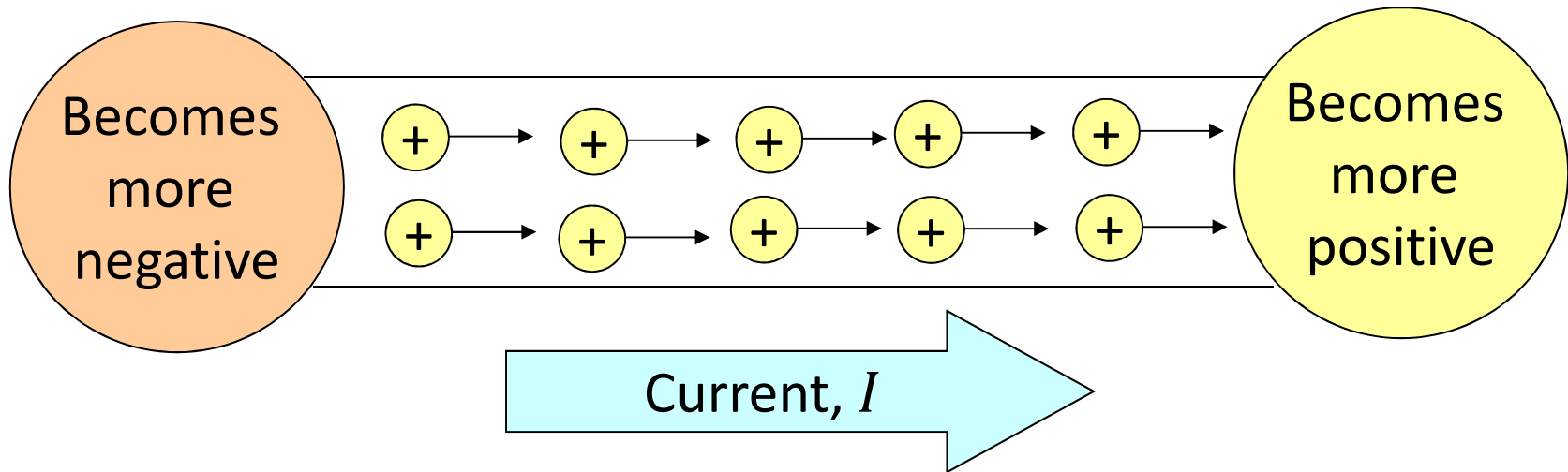
- Lecture 7:  $V(\vec{x}) = \frac{1}{4\pi\epsilon_0} \sum_i \frac{Q_i}{r_i} \rightarrow V(\vec{x}) = \frac{1}{4\pi\epsilon_0} \int \frac{dQ}{r}$
- Lecture 8:  $C = \frac{Q}{V}$  and  $U = \frac{1}{2} CV^2$
- Lecture 9:  $C_{\parallel} = C_1 + C_2$  and  $C_{series} = \left( \frac{1}{C_1} + \frac{1}{C_2} \right)^{-1}$
- But wait! There's more...
  - No  $\vec{E}$  field inside a conductor
  - Principle of superposition
  - Surface charge densities
  - $\vec{E}$  and  $V$  for example geometries
  - Work and energy
  - Dielectrics and  $\epsilon = \kappa \epsilon_0$
  - Lots of examples and clicker questions...

# Electrostatic Equilibrium

- No net motion of charge
- Insulators:
  - No free charges
- Conductors:
  - Charges are pushed by any electric field until their own electric field cancels the original one
  - The motion stops when charges accumulate at a surface
  - The net electric field in the conductor is zero
- What if charge is added to or removed from a surface as quickly as it accumulates?
  - The charge will continue to flow...
  - Not a state of electrostatic equilibrium

# Electric Current

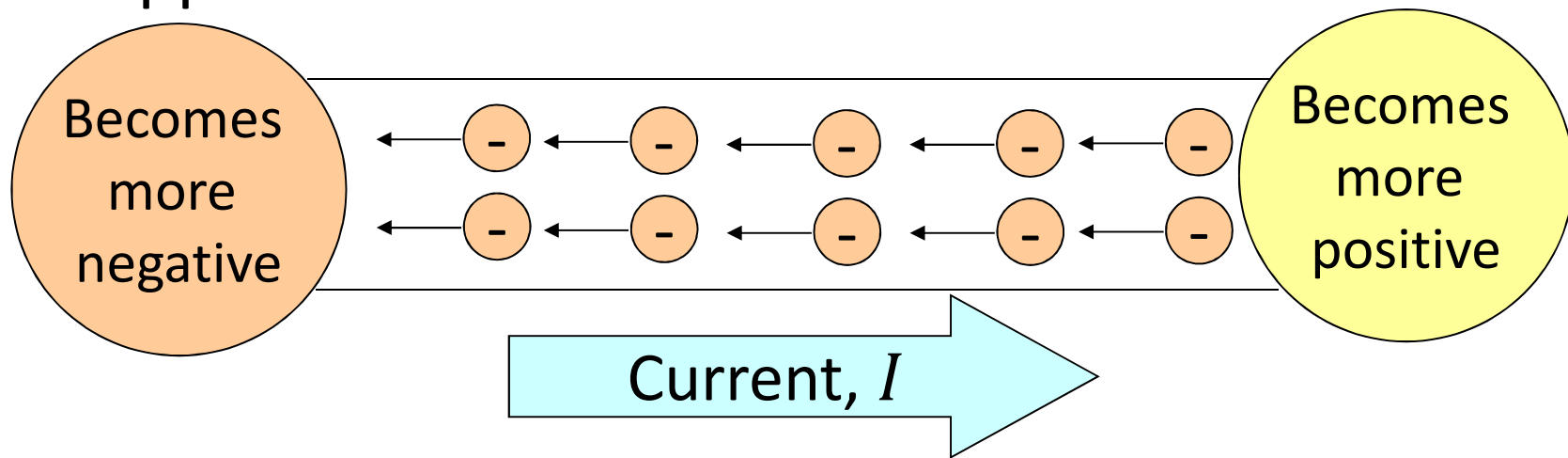
- When the motion of charge carriers are not restricted, they will flow.
- By convention, the direction of an electric current is in the direction that positive charge carriers move:





# Electric Current

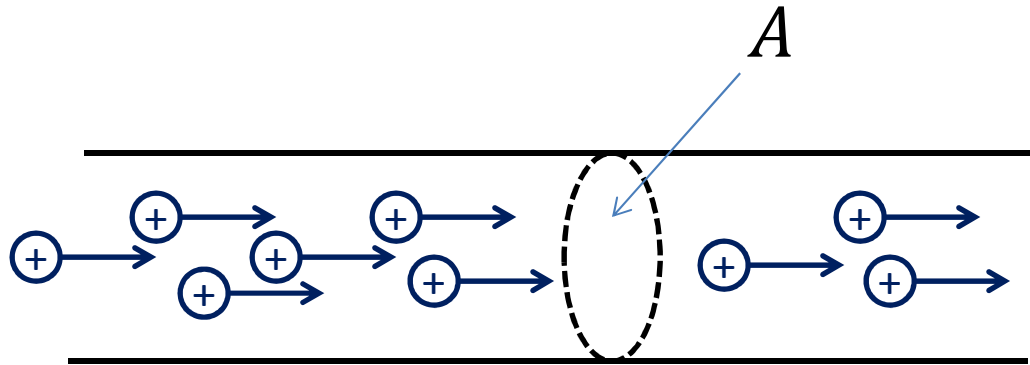
- If they are negatively charged, then the current is opposite their motion:



- In metals, the charge carriers are electrons
- In chemical solutions or ionized gasses, the charge carriers can be both positive and negative.

# Electric Current

- Electric current is the net positive charge moving across a surface per unit time:

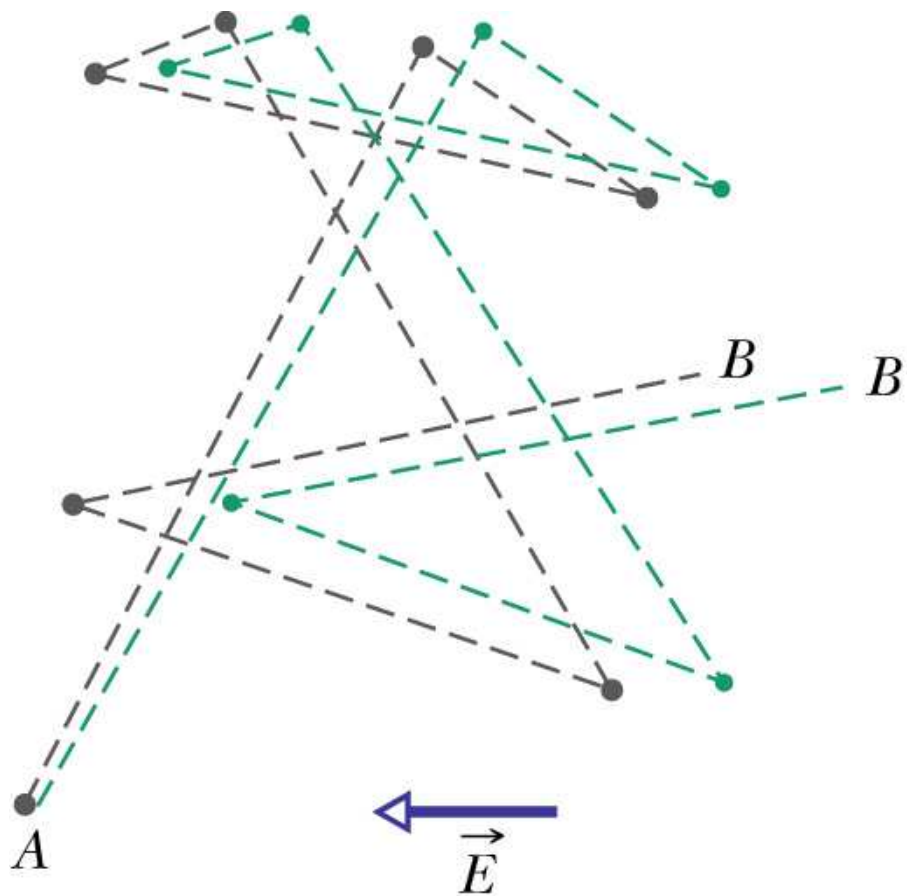


$$I = \frac{\Delta Q}{\Delta t}$$

$$\text{Units: } \textit{Amperes} = \frac{\textit{Coulombs}}{\textit{second}}$$

# Drift Velocity

- Motion of individual charges is usually not uniform:

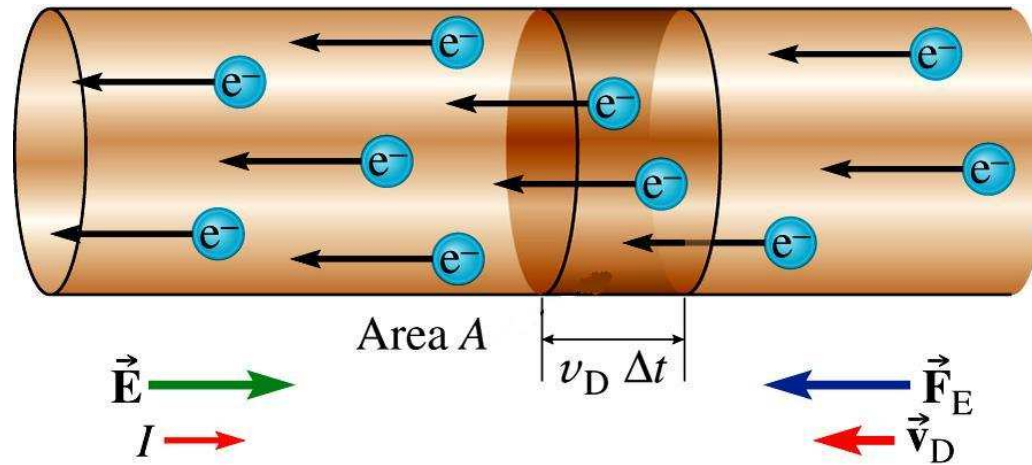


The average distance moved per unit time is the drift velocity:

$$v_D = \frac{\overline{\Delta x}}{\Delta t}$$

--- With  $\vec{E}$   
--- Without  $\vec{E}$

# Drift speed, total charge and current

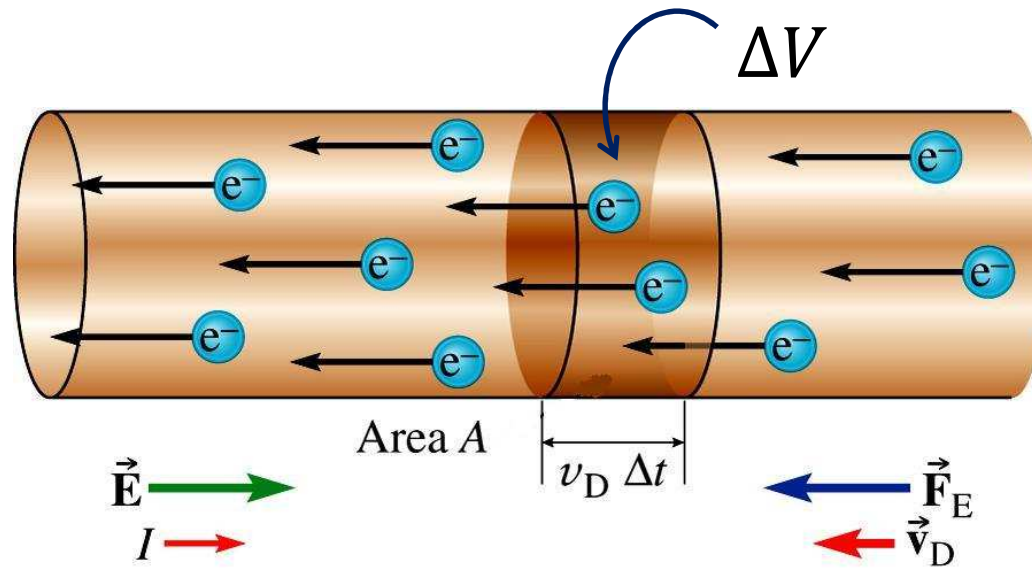


$$n = \frac{\text{\# of charge carriers}}{\text{unit volume}}$$

$q$  = charge of each carrier

$v_D$  = drift velocity

# Drift speed, total charge and current



$$\Delta Q = n q \Delta V$$

$$\Delta V = A v_D \Delta t$$

$$I = \frac{\Delta Q}{\Delta t} = \frac{n q A v_D \Delta t}{\Delta t} = n q A v_D$$

# Example

- What is the drift velocity in #12 AWG copper wire carrying 1 ampere of current?
  - What's the diameter of #12 AWG???
  - Google “wire gauge”... it's roughly 2 mm
  - How many charge carriers?
    - Assume one charge carrier per copper atom
    - How many copper atoms per unit volume?
    - How many copper atoms per unit mass?

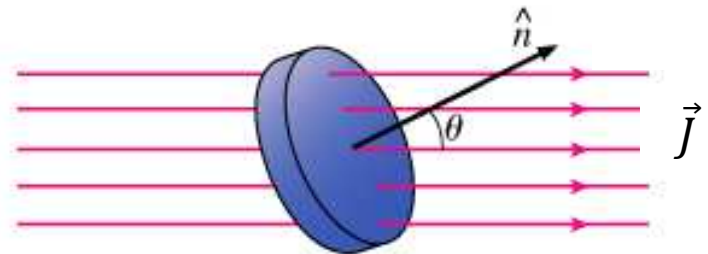
Atomic mass:  $m = 63.546 \text{ g/mol}$

Density of copper:  $\rho = 8.94 \text{ g/cm}^3$

# Current Density

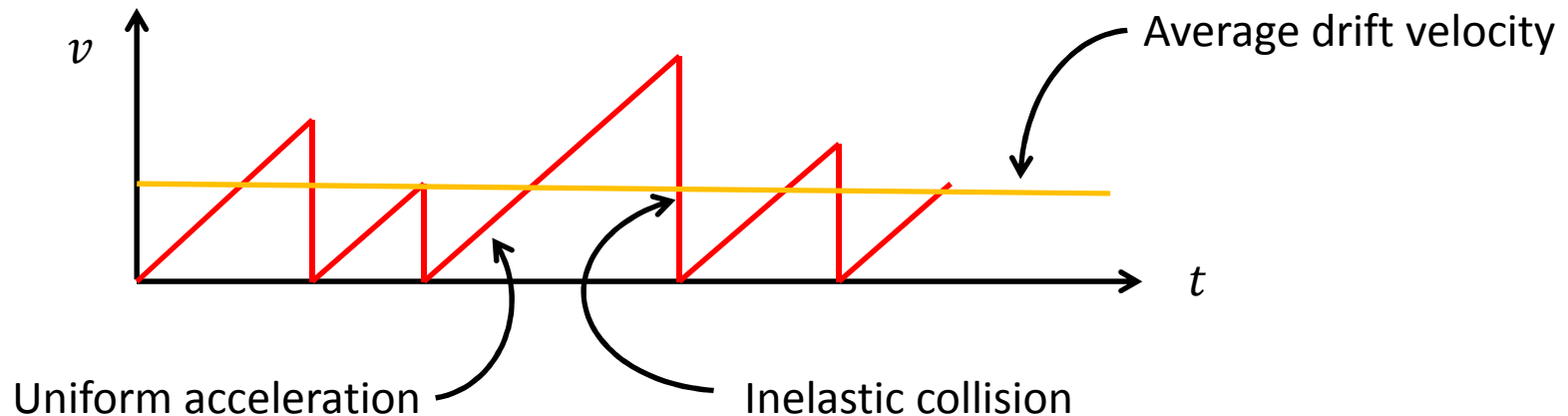
- The flow of charge might not be uniformly across a surface
  - The magnitude of the local current might change
  - The direction of the drift velocity could change
- Current:  $I = n q v_D A$
- Current density:  $\vec{J} = n q \vec{v}_D$
- They are related:

$$I = \int_S \vec{J} \cdot d\vec{A}$$



# Resistance

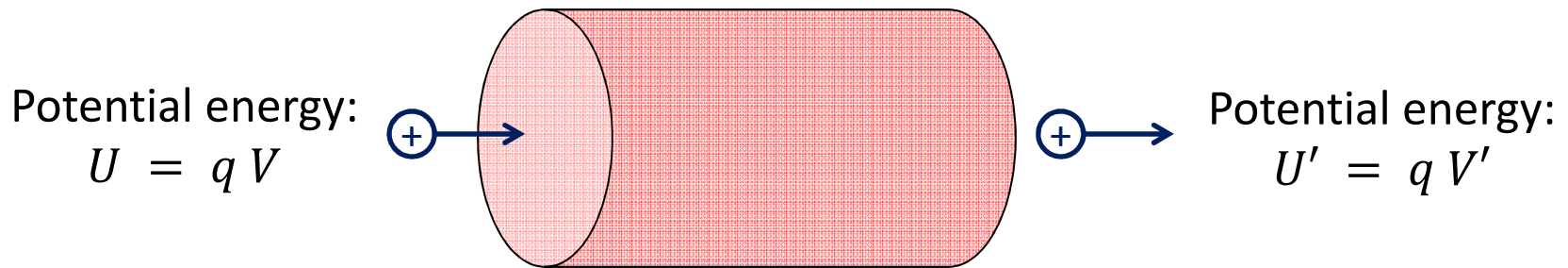
- Electrons in a metal do not accelerate indefinitely
  - They eventually hit an atom in the metal
  - The collision is inelastic and the electron loses all, or some of its energy
- Instantaneous vs average velocity:



- Resistance is a property of a material related to how rapidly charge carriers lose energy



# Resistance



- The greater the current, the more energy is transferred to the material through inelastic collisions.
- Electric potential difference:  $\Delta V \propto I$

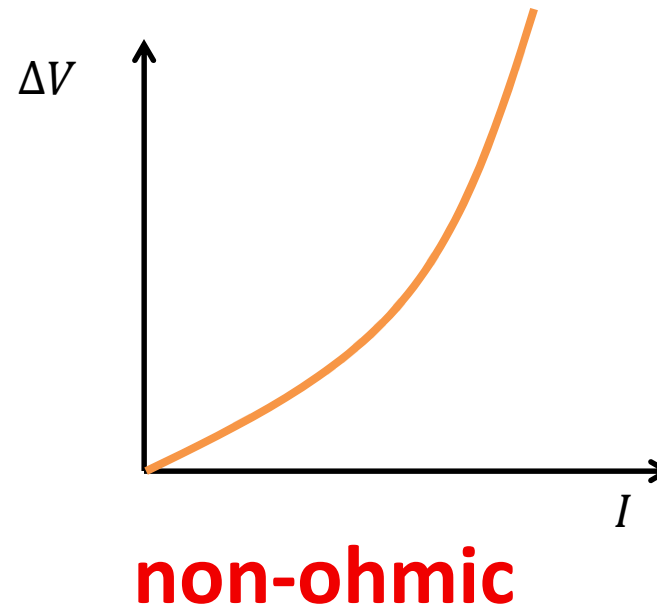
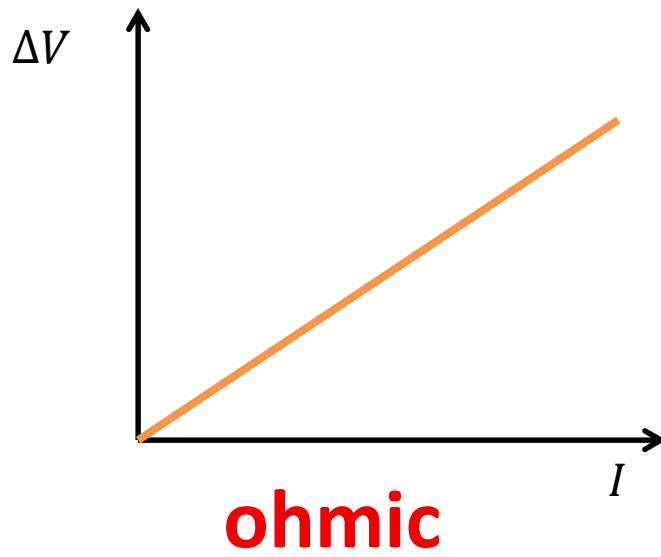
$$\Delta V = R I \quad \longrightarrow \quad R = \frac{\Delta V}{I}$$
$$\text{Ohms} = \frac{\text{Volts}}{\text{Ampere}}$$

# Ohm's Law

- Potential difference is proportional to current

$$\Delta V = I R$$

- This is usually a good approximation...

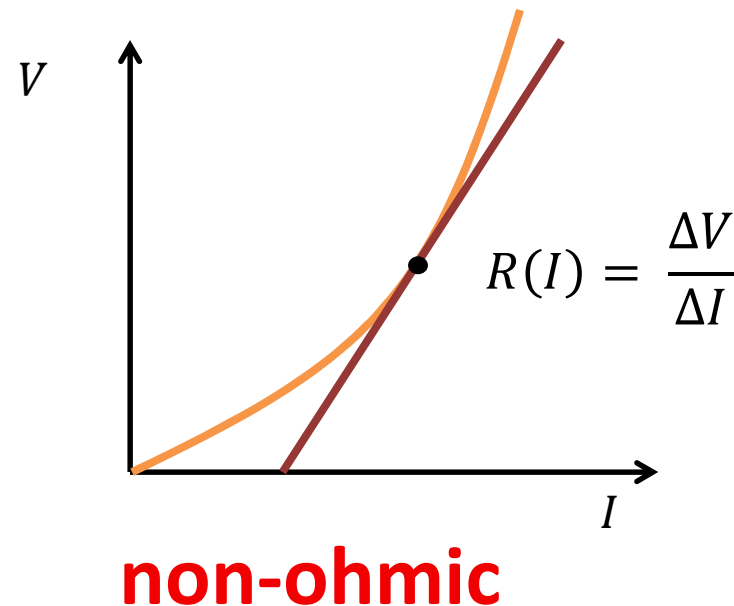
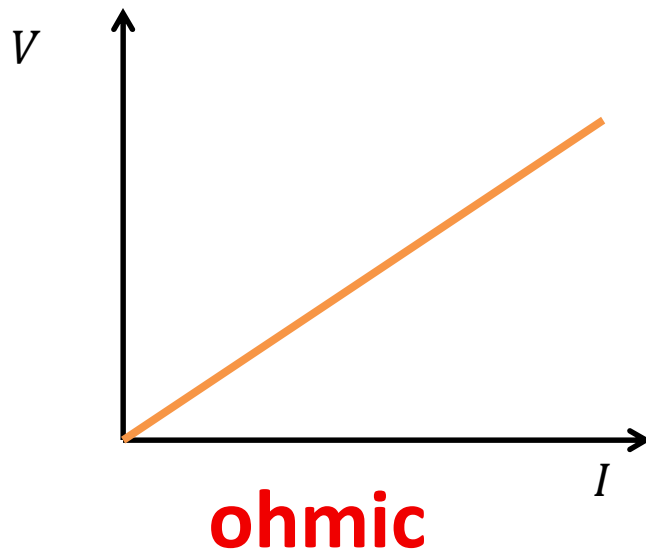


# Ohm's Law

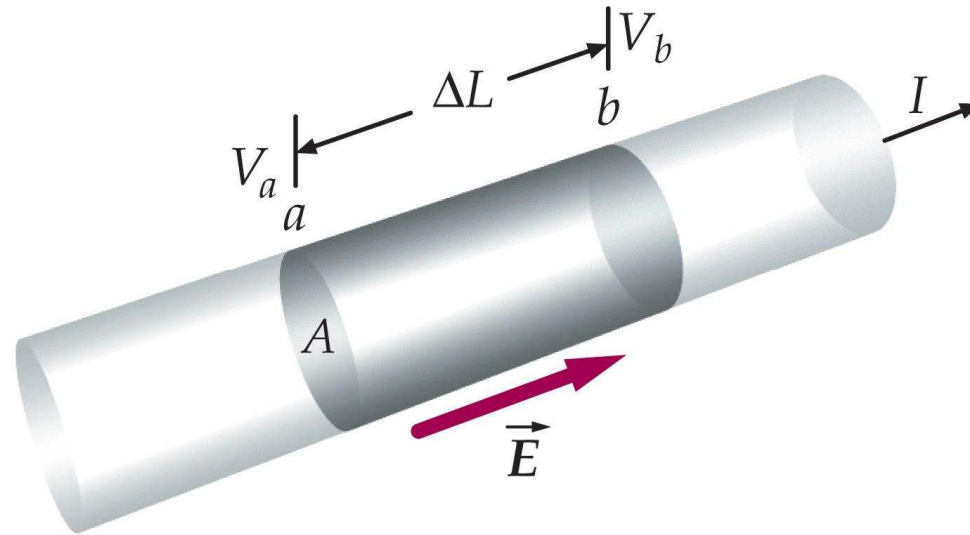
- Potential difference is proportional to current

$$\Delta V = I R$$

- This is usually a good approximation...



# Resistance depends on geometry



- Resistance is proportional to  $\Delta L$
- Resistance is inversely proportional to  $A$
- Resistivity,  $\rho$ , is independent of geometry

$$R = \frac{\rho \Delta L}{A}$$

# Resistivity Depends on Temperature

- In general, resistivity increases with temperature

$$\Delta\rho \propto \Delta T$$

- The temperature coefficient,  $\alpha$ , is defined as the fractional change in resistance:

$$\alpha = \frac{1}{\rho} \frac{\Delta\rho}{\Delta T}$$

- Resistivity and the temperature coefficient are usually given for a particular reference temperature (for example, 20 °C)

# Resistivities and Temperature Coefficients

Material	Resistivity, $\rho$ ( $\Omega \cdot m$ )	Temp. coeff., $\alpha$ ( $K^{-1}$ )
Ag	$1.6 \times 10^{-8}$	$3.8 \times 10^{-3}$
Cu	$1.7 \times 10^{-8}$	$3.9 \times 10^{-3}$
W	$5.5 \times 10^{-8}$	$4.5 \times 10^{-3}$
Si	640	$-7.5 \times 10^{-2}$
Si, n-type	$8.7 \times 10^{-4}$	
Si, p-type	$2.8 \times 10^{-3}$	
glass	$10^{10} - 10^{14}$	

# Temperature Dependence

$$\rho - \rho_0 = \rho_0 \alpha (T - T_0)$$
$$\rho(T) = \rho_0 (1 + \alpha (T - T_0))$$

- Also true for resistance:  $R = \rho L / A$

$$R - R_0 = R_0 \alpha (T - T_0)$$
$$R(T) = R_0 (1 + \alpha (T - T_0))$$

# Example

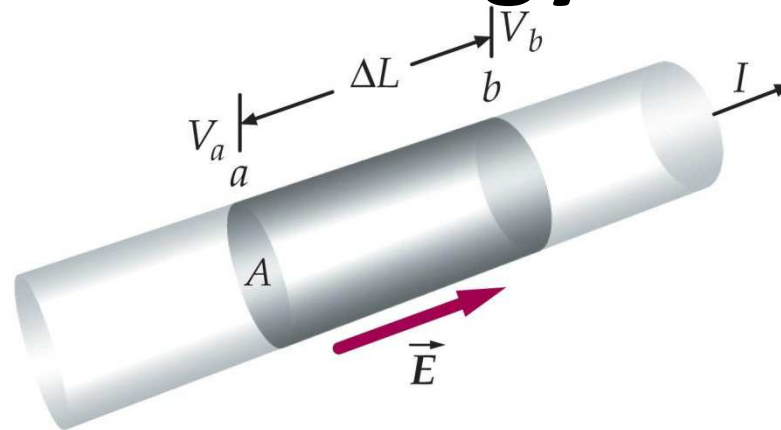
- What is the resistance of a 10 cm long Tungsten wire with a diameter of 0.2 mm at 20 °C and at 3000 K?

$$\rho_0 = 5.5 \times 10^{-8} \Omega \cdot m$$

$$\alpha = 4.5 \times 10^{-3} K^{-1}$$



# Rate of Energy Loss



- Charges moving through a resistor lose energy

$$\Delta U = q \Delta V = q I R$$

- Total energy lost per unit time:

$$P = n q A v_D \times I R$$

$$P = I^2 R$$

- Electric potential energy is converted into heat.

# Clicker Question

- The resistance across the human body is approximately  $2\text{ k}\Omega$
- If it takes only  $50\text{ mA}$  of current to kill a human, what voltage could be lethal?

(a) 0.1 Volts

(b) 1 Volt

(c) 10 Volts

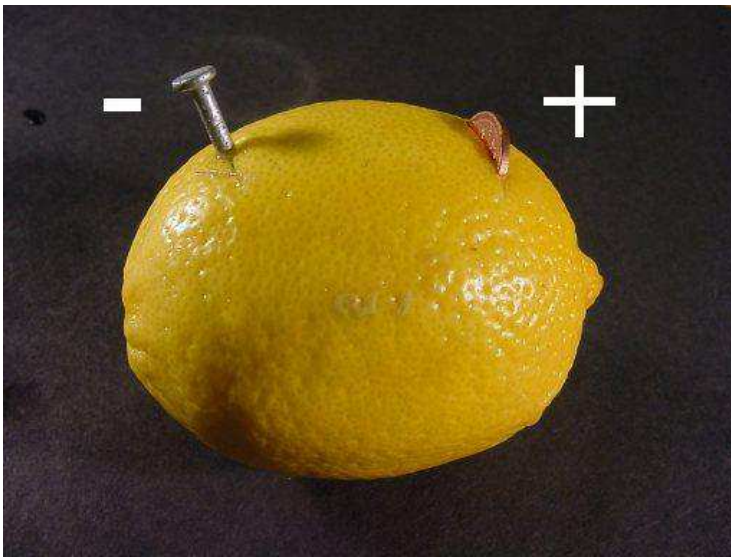
(d) 100 Volts

(e) 1000 Volts



# Electric Current

- A chemical battery is a *source* of electric potential
  - The chemical reaction creates a potential difference across the poles:



The chemical reaction maintains a constant potential difference.

Positive charges at the + end have a greater electric potential than positive charges at the – end.

There must be an electric field between the poles.

If free charges were present, they would be accelerated by the electric field – the field does work on the charges.

Their potential energy decreases as they move towards the – pole.