

6.2 Average Values

For any statistical treatment, the average of a variable is the probability that value i occurs summed over all possibilities.

For example, the probability for $S_z = \hbar/2$ is 0.7 and the probability for $S_z = -\hbar/2$ is 0.3
 $\Rightarrow \bar{S}_z = 0.7 \hbar/2 + 0.3(-\hbar/2) = 0.2\hbar$

For our "system", the probability to be in state s is

$$P(s) = \frac{1}{Z} e^{-\beta E(s)} \quad \beta = \frac{1}{k_B T} \text{ (this is standard notation)}$$

$$\bar{E} = \sum_s E(s) P(s) = \frac{1}{Z} \sum_s E(s) e^{-\beta E(s)}$$

In general if there is a quantity $X(s)$ [X might be spin, KE, position ...]

$$\bar{X} = \sum_s X(s) P(s) = \frac{1}{Z} \sum_s X(s) e^{-\beta E(s)}$$

Suppose you have N identical objects make up your system (objects are independent and distinguishable)
The total energy is $N \times$ average energy of 1

$$U = N \bar{E} \quad (\text{Note trick! Can get } U \text{ by doing 1 object.})$$

For quantum system how does U behave at low T ?

$$\bar{E} = E_1 P(1) + E_2 P(2) \approx E_2 \frac{\Omega_2}{\Omega_1} e^{-E_2/k_B T}$$

$$\Rightarrow U = N E_2 \frac{\Omega_2}{\Omega_1} e^{-E_2/k_B T}$$

$$\text{The heat capacity } C_V / N k_B = \frac{1}{N k_B} \left(\frac{\partial U}{\partial T} \right)_N = \left(\frac{E_2}{k_B T} \right)^2 \frac{\Omega_2}{\Omega_1} e^{-E_2/k_B T}$$

$$C_V \rightarrow 0 \text{ as } T \rightarrow 0$$

Prob 6.16 Prove $\bar{E} = -\frac{1}{Z} \frac{\partial Z}{\partial \beta} = -\frac{\partial}{\partial \beta} \ln Z$

$$Z = \sum_s e^{-\beta E(s)} \Rightarrow -\frac{\partial Z}{\partial \beta} = \sum_s E(s) e^{-\beta E(s)} = \bar{E} \cdot Z$$

Prob 6.17 Standard deviation

a) $\bar{E} = 3 \text{ eV}$ $\Delta E_1 = 0 - 3 \text{ eV} = -3 \text{ eV}$, $\Delta E_2 = 1 \text{ eV}$, $\Delta E_3 = 4 \text{ eV}$

b) $\overline{\Delta E} = \frac{2}{5}(-3 \text{ eV}) + \frac{2}{5} 1 \text{ eV} + \frac{1}{5} 4 \text{ eV} = 0 \text{ ?!!}$

$$\overline{\Delta E^2} = \frac{2}{5} 9 \text{ eV}^2 + \frac{2}{5} 1 \text{ eV}^2 + \frac{1}{5} 16 \text{ eV}^2 = \frac{36}{5} \text{ eV}^2$$

$$\sigma_E \equiv \sqrt{\overline{\Delta E^2}} = \frac{6}{\sqrt{5}} \text{ eV} \approx 2.7 \text{ eV} \text{ reasonable?}$$

c) $\sigma_E^2 = \sum_i (E_i - \bar{E})^2 P_i = \sum_i (E_i^2 - 2E_i \bar{E} + \bar{E}^2) P_i$
 $= \sum_i E_i^2 P_i - 2\bar{E} \sum_i E_i P_i + \bar{E}^2 \sum_i P_i$
 $= \overline{E^2} - 2\bar{E}^2 + \bar{E}^2 = \overline{E^2} - \bar{E}^2$

d) $\overline{E^2} = 0 \text{ eV}^2 \frac{2}{5} + 16 \text{ eV}^2 \frac{2}{5} + 49 \text{ eV}^2 \frac{1}{5} = \frac{81}{5} \text{ eV}^2$
 $\overline{E^2} - \bar{E}^2 = \frac{81}{5} \text{ eV}^2 - 9 \text{ eV}^2 = \frac{36}{5} \text{ eV}^2 \checkmark$

Prob 6.18 Prove $\overline{E^2} = \frac{1}{Z} \frac{\partial^2 Z}{\partial \beta^2}$

$$Z = \sum_s e^{-\beta E(s)} \quad \frac{\partial^2 Z}{\partial \beta^2} = \sum_s E(s)^2 e^{-\beta E(s)} = Z \overline{E^2}$$

Now compute $\sigma_E^2 = \overline{E^2} - \bar{E}^2 = \frac{1}{Z} \frac{\partial^2 Z}{\partial \beta^2} - \left(\frac{\partial Z}{\partial \beta} \right)^2$

Remember $\bar{E} = -\frac{1}{Z} \frac{\partial Z}{\partial \beta}$ $\frac{\partial \bar{E}}{\partial \beta} = \frac{\partial}{\partial \beta} \left(-\frac{1}{Z} \right) \frac{\partial Z}{\partial \beta} - \frac{1}{Z} \frac{\partial^2 Z}{\partial \beta^2}$
 $= -\frac{1}{Z^2} \left(\frac{\partial Z}{\partial \beta} \right)^2 - \frac{1}{Z} \frac{\partial^2 Z}{\partial \beta^2}$

$$\Rightarrow \sigma_E^2 = -\frac{\partial \bar{E}}{\partial \beta}$$

Want answer in terms of $\frac{\partial \bar{E}}{\partial T} = \frac{\partial \bar{E}}{\partial \beta} \frac{\partial \beta}{\partial T} = -\frac{1}{k_B T^2} \frac{\partial \bar{E}}{\partial \beta} = \frac{\sigma_E^2}{k_B T^2}$

$$\Rightarrow \sigma_E = k_B T \sqrt{\frac{\partial \bar{E}}{\partial T} k_B} = k_B T \sqrt{C/k_B} \quad C = \text{heat capacity!}$$

What does this mean for energy fluctuation of macroscopic system? $\sigma_E \propto \sqrt{N}$, but $U \propto N$

$\Rightarrow$ relative size of fluctuation $\sigma_E/U \propto 1/\sqrt{N}$

For low T system $C = \frac{\sum_i k_B E_i^2}{\sum_i \frac{1}{k_B T^2} e^{-E_i/k_B T}} \Rightarrow \sigma_E = E_2 \sqrt{\frac{R_2}{R_1}} e^{-\frac{E_2}{2k_B T}} \rightarrow 0$
 as $T \rightarrow 0$

Does this make sense?

Paramagnetism (revisit 3.3)

$$1 = \uparrow \quad E_1 = -\mu B \quad 2 = \downarrow \quad E_2 = \mu B$$

$$Z = e^{\mu B \beta} + e^{-\mu B \beta} = 2 \cosh(\beta \mu B)$$

$$P_1 = \frac{e^{\mu B \beta}}{Z} = \frac{e^{\mu B \beta}}{2 \cosh(\beta \mu B)} \quad P_2 = \frac{e^{-\mu B \beta}}{2 \cosh(\beta \mu B)} \quad \checkmark$$

$$\bar{E} = -\mu B P_1 + \mu B P_2 = -\mu B (P_1 - P_2) = -\mu B \frac{\sinh(\beta \mu B)}{\cosh(\beta \mu B)} \\ = -\mu B \tanh(\beta \mu B) \quad \checkmark$$

$$U = N \bar{E} = -N \mu B \tanh(\beta \mu B) \quad \checkmark$$

$$\bar{\mu}_z = \mu P_1 - \mu P_2 = \mu (P_1 - P_2) = \mu \tanh(\beta \mu B)$$

$$M = N \bar{\mu}_z = N \mu \tanh(\beta \mu B) \quad \checkmark$$

All results reproduced with much less effort!

Rotation of diatomic molecules (simplified version)

Remember from classical mechanics $E = \frac{L^2}{2I}$ for rotational KE

$$\Rightarrow E = l(l+1) \epsilon \quad \epsilon = \frac{\hbar^2}{2I}$$

$$l = 0, 1, 2, 3, \dots$$

How many states for each l ?

$$l=0 \quad m=0 \quad 1 = 2l+1$$

$$l=1 \quad m=-1, 0, 1 \quad 3 = 2l+1$$

$$l=2 \quad m=-2, -1, 0, 1, 2 \quad 5 = 2l+1$$

$$\Rightarrow Z_{\text{rot}} = \sum_{l=0}^{\infty} (2l+1) e^{-l(l+1)\epsilon/k_B T}$$

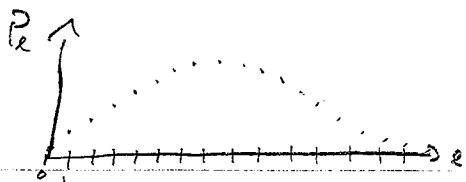
If $\epsilon/k_B T > 1$ then only a few levels will contribute

From prev. pages

$$U \approx N Z \epsilon \cdot 3 e^{-2\epsilon/k_B T} \quad \text{and} \quad C_V/Nk_B \approx \left(\frac{2\epsilon}{k_B T}\right)^2 3 e^{-2\epsilon/k_B T}$$

At higher T , $\epsilon/k_B T \ll 1$, you can use a different approximation. A huge number of levels contribute to the sum in Z so convert it to integrat.

High T



$$Z_{\text{rot}} \approx \int_0^{\infty} (2l+1) e^{-l(l+1)\epsilon/k_B T} dl = -\frac{k_B T}{\epsilon} \int_0^{\infty} \left[\frac{d}{dl} e^{-(l^2+l)\epsilon/k_B T} \right] dl$$

$$\Rightarrow Z_{\text{rot}} \approx \frac{k_B T}{\epsilon} = \frac{1}{\beta \epsilon}$$

$$\bar{E}_{\text{rot}} = -\frac{1}{Z_{\text{rot}}} \frac{\partial Z_{\text{rot}}}{\partial \beta} = -\beta \epsilon \frac{\partial}{\partial \beta} \left(\frac{1}{\beta \epsilon} \right) = \frac{1}{\beta} = k_B T !!$$

Equipartition Thm
rears its ugly head!

Couple problems with rotation of diatomic molecules:

1) If the electron part of wave function has nonzero $\bar{J}$, then l not correct. Need total angular momentum.

2) Identical atoms the wave function needs appropriate symmetry.

2 spin 0 atoms only even l allowed

At high T $\Rightarrow Z_{\text{rot}} \approx \frac{1}{2} \frac{k_B T}{\epsilon}$ however $\bar{E}_{\text{rot}} = k_B T$ still.

Prob 6.24
Pg 236

$$\epsilon = 0.00018 \text{ eV} \Rightarrow \frac{\epsilon}{k_B T} = \frac{0.00018 \text{ eV} \cdot 1.6 \times 10^{19} \text{ J/eV}}{1.38 \times 10^{-23} \text{ J/K} \cdot 300 \text{ K}} = 6.96 \times 10^{-3}$$

$$\Rightarrow Z \approx \frac{1}{2} \frac{k_B T}{\epsilon} = 72$$

Prob 6.27
Pg 237

kT/ϵ	Z_{exact}
0	1
1	1.418
2	2.370
3	3.357
4	4.357
6	6.345
10	10.340
15	15.338
30	30.336
60	60.334

From excel

$$Z_{\text{exact}} \approx \frac{k_B T}{\epsilon} + \frac{1}{3} = \frac{1}{\beta \epsilon} + \frac{1}{3}$$

$\Rightarrow$ the $\frac{1}{3}$ doesn't mess up $\bar{E}_{\text{rot}}$

$$\bar{E}_{\text{rot}} = -\frac{1}{\frac{1}{\beta \epsilon} + \frac{1}{3}} \frac{\partial}{\partial \beta} \left(\frac{1}{\beta \epsilon} + \frac{1}{3} \right) = \frac{k_B T}{1 + \frac{\epsilon}{3k_B T}}$$

To see where the $\frac{1}{3}$ comes from
note $\int_0^{\infty} F(x) dx \approx \frac{5}{12} F_0 + \frac{13}{12} F_1 + F_2 + F_3 + \dots$

$$\begin{aligned} \Rightarrow F_0 + F_1 + F_2 + \dots &\approx \int_0^{\infty} F(x) dx + \frac{7}{12} F_0 - \frac{1}{12} F_1 \\ &= \frac{k_B T}{\epsilon} + \frac{7}{12} \left(1 - \frac{1}{12} 3e^{-2\epsilon/k_B T} \right) \\ &= \frac{k_B T}{\epsilon} + \frac{7}{12} - \frac{3}{12} = \frac{k_B T}{\epsilon} + \frac{1}{3} ! \end{aligned}$$