Some HF Facts

HF is **not** just a "weak acid"

HF is used in freshman chemistry as a classic example of a weak acid (pK_a = 3.17, as contrasted with its apparently close relative HCl, pK_a = -3)

 But it is also hideously toxic, in a way that most people will not guess or suspect.

HF or HF_(aq) exposure

• HF_(in cells)
$$\frac{}{}$$
 $\frac{}{}$ H+_(aq) + F-_(aq) acid thing

Summary: the Ca²⁺ and Mg²⁺ are sucked almost instantly out of your cells, sera, and interstitial fluids. Your tissues and metabolic pathways are messed with in a horrifying and nearly irreversible manner.

rxn with Ca²⁺ and Mg²⁺ in cells is rapid, not much reverse rxn

CaF₂ and MgF₂ (insoluble)

HF First Aid

 HF first aid treatment is not limited to simply washing it off the skin or out of the eyes

- HF penetrates (is absorbed) very rapidly through the skin and mucous membranes
- Deep tissue necrosis is fast.
 Amputation can be necessary to save life.



This is a small HF burn, from low concentration HF.

The time before onset of pain is concentration dependent

• 0 - 20% HF

up to 24 h

• 20% - 50% HF 1 - 8 h

• 50% - 70% HF <1 h

• >70% HF

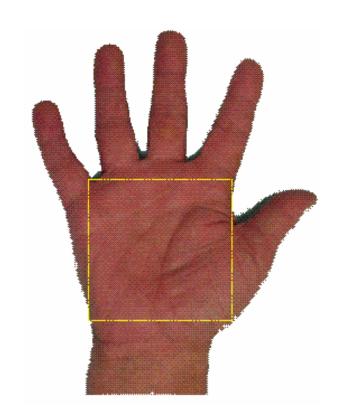
immediate





PAIN is a very important symptom, but it's better not to wait for symptoms, in general.

First Aid must be immediate



- Aqueous 50% HF on 1% of the body surface area sometimes fatal <u>even if</u> <u>immediate appropriate</u> first aid is rendered.
- 1% of BSA (<u>b</u>ody <u>s</u>urface <u>a</u>rea) is approx equal to palm surface or a bit more.
- 5% of BSA exposed to any concentration HF has similar effect.
- As little as 7 mL anhydrous HF can bind all of the free calcium in a normal sized adult male.