Project on Brine-Based Remediation of DNAPL-Contaminated Subsurface Systems C.T. Miller and R.M. McLaughlin University of North Carolina

Motivation

- Multiphase systems result from common sources of contamination
- Remediation of such contaminated systems has proven especially difficult
- A wide variety of schemes have been investigated to accomplish such a remediation
- Brine-based remediation methods motivate this project

Characteristics of Behavior

- NAPLs leave a state of residual saturation in media through which they pass
- NAPLs follow a complex pattern of flow, which is importantly influenced by media heterogeneity
- LNAPLs accumulate on the top of the water table
- DNAPLs can sink below the water
- NAPLs often reach stable configurations of locally high saturations known as pools
- NAPLs are usually sparingly soluble and DNAPL contaminants usually degrade slowly---thus are long lived in the environment

DNAPL Behavior in Heterogeneous Porous Media



Current Remediation Approaches and Limitations

- Pump-and-Treat
- Vapor-Phase Extraction
- Air Sparging
- Cosolvent and Surfactant Flushing
- Thermal Processes
- In Situ Biodegradation

Pump-and-Treat



Federal Remediation Technology: http://www.frtr.gov/matrix2/section4/D01-4-48.html

Vapor Extraction



Federal Remediation Technology: http://www.frtr.gov/matrix2/section4/D01-4-7.gif

Air Sparging



Federal Remediation Technology: http://www.frtr.gov/matrix2/section4/D01-4-34.gif

Cosolvent and Surfactant Flushing



Low Permeability Zone

Federal Remediation Technology: http://www.frtr.gov/matrix2/section4/D01-4-6.gif

Thermal Processes



Federal Remediation Technology: http://www.frtr.gov/matrix2/section4/D01-4-38.gif

In Situ Biodegradation



Regional Aquifer

Federal Remediation Technology: http://www.frtr.gov/matrix2/section4/D01-4-2.gif

Summary of Current Approaches

- Mass transfer limitations are important for all technologies that do not mobilize the NAPL---leading to long clean-up times
- Technologies that do mobilize the NAPL phase suffer from uncontrolled mobilization that can contaminate previously clean portions of a system
- Invasive techniques can be prohibitively expensive
- In situ removal is a difficult consideration, but effective remediation methods also must solve the waste stream treatment problem
- No silver bullet: no method will be universally the best choice and economics of restoration will be site dependent

Objectives of Effective Remediation

- Remove source zone of long-lived contaminants
- Do not rely on technologies that can be limited by a slow mass transfer process
- Avoid technologies that can spread a contaminant to previously clean portions of a system
- Target approaches that can reduce a sufficient fraction of the source mass in a relatively short period of time
- Consider technologies that have manageable aboveground treatment requirements and allow reuse of flushing solutions

Brine-Based Remediation Methods

- Manipulate density of aqueous phase to ensure NAPL mobilization is controlled
- Affect balance of forces to free NAPL trapped by capillary forces
- Capture mobilized NAPL as a free phase from the top of the relatively dense aqueous phase
- Use surfactant flushing and vapor extraction to further reduce NAPL residual
- Recycle and recover flushing solutions as appropriate
- Treat and separate waste stream with above-ground unit processes

Two-Dimensional Unsaturated Downward Vertical Displacement of TCE

2001]



•21-cm x 21-cm two-dimensional cell Pooled TCE established •TCE dyed with Oil Red O for visualization •Established bottom brine layer •Drained to unsaturated conditions •0.3 pore-volume downward flush with mixture of sulfosuccinate surfantants •Measured 80.0% TCE removal, no visible pools •Reference: Hill et. al. [ES&T, 35(14),

Three-Dimensional Unsaturated Downward Vertical Displacement of TCE



•22-cm x 24-cm x 16-cm threedimensional cell

- •Pooled TCE established in heterogeneous media
- •TCE dyed with Oil Red O for visualization
- •Established bottom brine layer
- Drained to unsaturated conditions
- •0.2 pore-volume downward flush with mixture of sulfosuccinate surfantants
- •Measured 63.4% TCE removal, no visible pools

•Reference: Hill et. al. [ES&T, 35(14), 2001]

Three-Dimensional Density-Motivated Mobilization Experiments

Setup



Three-Dimensional Density-Motivated Mobilization Experiments

Cumulative Recovery of TCE



Three-Dimensional Density-Motivated Mobilization Experiments

Properties and Recovery

	No.	Pore		TCE Recovery (%)			Mass
	Extraction Wells	Volume (L)	Surfactant (PV)	Well Extraction	Vapor Extraction	Soil Extraction	Balance (%)
Experiment #1	1	1.3	1.8	76.5	n/a	14.7	91.2
Experiment #2	1	1.5	2.6	83.5	5.7	1.0	90.2
Experiment #3	3	1.4	5.3	86.2	8.2	0.4	94.8

Three-Dimensional Unsaturated Downward Vertical Displacement of TCE



•22-cm x 24-cm x 16-cm threedimensional cell

•Pooled TCE established in heterogeneous media

•TCE dyed with Oil Red O for visualization

•Established bottom brine layer

•Drained to unsaturated conditions

•3.2 pore-volume downward flush with mixture of sulfosuccinate surfantants followed by vapor extraction

•Measured 99% TCE removal of recovered TCE

•Reference: Johnson et. al. [ES&T, 38(19), 2004]

Open Issues

- Scale up
- Brine density control and recovery
- Surfactant selection
- Geochemical stability
- Waste-stream separation and process treatment design and pilot testing
- Mathematical model development and application
- Development of optimal design strategies
- Economical analysis

Dover National Test Site

Field Operations Office/Lab

Weather Station

Jet Grout Barrier

Storage Tent

Test Cells 2 & 3

Dover National Test Site



Depth to the water table is approximately 28 feet.

Aquifer depth is approximately 12 feet.

Test cells are double-walled sheet piles driven into the subsurface.

Sheet piles are keyed into a confining aquitard approximately 45 feet below the surface.

Technical Approach



Brine Density



Brine Viscosity



Formation of Brine Layer



- Three-dimensional experiment
- Dover-like sand
- Brine injected from bottom
- Density monitored throughout the system and with time
- Density of the brine layer exceeds the density of PCE after about 33 hours

Diffusion of Brine



- Diffusion of brine aboutsharp interface observedin space and time
- Density of 1.7 g/mL corresponds to a TDS of 900,000 mg/L
- PCE density 1.62 g/mL corresponded to a TDS of 780,000 mg/L
- Brine barrier is stable and long-lived in presence of diffusion alone

Recovery of Brine

igodol



- Brine removed bydrainage from upper,then lower, ports
- After drainage, horizontal flushing performed
- Water table reduced further as flushing continued

Effluent Density During Recovery



- Three-dimensional cell
- Dover-like sand
 - Drained first from top of brine layer
 - Drainage from within brine layer
- Horizontal flushing to observe brine residual removal

Effluent Brine Concentration During Recovery



- Three-dimensional cell
- Dover-like sand
- Drained first from top of brine layer
- Drainage from within brine layer
- Horizontal flushing to observe brine residual removal

Project

- Focus on brine emplacement and removal aspects of brine-based technology
- Investigate effects of viscous and gravity instabilities
- Investigate effects of media heterogeneity and learn about homogenization-based upscaling methods